

Asbury Graphite Mills, Inc. Cummings – Moore Graphite Co. Anthracite Industries Southwestern Graphite **Asbury Graphite of California Asbury – Wilkinson** Asbury Graphite & Carbons NL B.V. Graphitos Mexicanos de Asbury, S.A. de C.V.

PO Box 144, 405 Old Main St. Asbury, NJ 08802	908-537-2155
1646 N. Green Ave. Detroit, MI 48209	313-841-1615
PO Box 112, Sunbury, PA 17801	570-286-2176
PO Box 876, 2564 Hwy 12 DeQuincy, LA 70633	337-786-5905
2855 Franklin Canyon Rd. Rodeo, CA 94572	510-799-3636
1115 Sutton Drive Burlington, ON, L7L 5Z8 Canada	905-332-0862
Fregatweg 46 B-C, Maastricht 6222 NZ Netherlands	+31437600610
Blvd José Maria Morelos No.389 Nte, Hermosillo 83148 Mexico	526622678598

Safety Data Sheet

Section 1 – Identification of the Substance / Preparation, and of the Company

1.1: Product Identifier

Trade Name: Synthetic Graphite 99%+ Carbon Grade: TC307

REACH Registration Number: 01-2119486977-12-0027 Substance Name: Graphite, CAS 7782-42-5

EC Number: 231-955-3

1.2: Indentified uses of the substance or mixtures

- 1.2.1 Uses: Inorganic source of carbon, filler, thermal additive, re-carburizer, casting powders, drilling fluids, plastic additive, rubber additive, tint/pigment, lubricant, chemically resistant additive, EMF absorber, , general inert filler-additive.
- 1.2.2 Uses Advised Against: For industrial use only, not for food, drug, or cosmetic applications.

1.3: Supplier Information

Company/Manufacturer: Asbury Carbons, Inc. Telephone: 908-537-2155 PO Box 144, 405 Old Main Street Telefax: 908-723-2908

> Asbury, NJ 08802 Preparer: AVT

Email Address: albert@asbury.com

Date Prepared: 5/5/2015

1.4: Emergency Telephone Number 1-800-255-3924

Section 2: Hazards Identification

2.1: Classification of substance

Synthetic Graphite is not a hazardous substance

2.2: Label Elements

Synthetic Graphite is not a hazardous substance, no label elements are required

2.3: Other hazards: None known















Section 3 – Composition/Information on Ingredients:

Chemical Composition: Carbon variety Graphite 99+% (balance is inert ash)

CAS # 7782-42-5 EC # 231-955-3 Molecular Weight: 12.0

Section 4 - First Aid Measures

4.1.1	Remove patient to particulate-free environment. Wear approved dust mask to avoid breathing
Inhalation	dust. Seek medical attention if irritation persists.
4.1.2 Skin	Wash with mild soap and warm water: Graphite is non-staining to skin and is not a chemical
Contact	irritant.
4.1.3 Eye	Rinse with tepid water until eyes are clear of particulates. Seek medical attention if irritation
Contact	persists.
4.1.4	Get immediate medical attention. Do not induce vomiting unless directed by medical personnel.
Ingestion	Synthetic graphite is not known to be toxic by ingestion. However, ingestion may cause digestive system blockage.

4.2 Most important symptoms and effects, both acute and delayed: No Data Available

4.3 Indication of any immediate medical attention and special treatment needed: If patient exhibits shortness of breath, choking, powder inundated eyes or mouth; immediate medical attention may be required.

Section 5 - Fire Fighting Measures

Graphite is not flammable under normal conditions					
5.1 Extinguishing Media	5.1 Extinguishing Media Dry chemical extinguisher, water, sand, limestone powder,				
5.2 Special Hazards	At temperatures above 1500 C, graphite reacts with substances containing oxygen, including water and carbon dioxide. In case of intensely hot fire events, use sand to cover and isolate graphite.				
Products of Combustion:	Carbon dioxide, CO2, carbon monoxide, CO.				
5.3 Advice for Fire Fighters: Use self contained air pack, gloves, safety goggles					
5.4 Additional Information: USA NFP Rating 110					

Section 6 - Accidental Release Measures

	Wear approved dust mask, safety goggles, and conventional work gloves.				
Methods for Cleaning Up:	Conventional Sweep or vacuum. Avoid creating dusting conditions				
6.1 Personal precautions , protective equipment and emergency procedures					

- 6.1.1 For non-emergency personnel: Wear approved dust mask, safety goggles, and conventional work gloves. Use conventional cleanup techniques and avoid creating dust. Vacuum is preferred over sweeping. Be cautious of slip hazard on wet or dry pedestrian surfaces. Wear a dust mask/respirator to reduce the change of inhaled dust. Graphite is electrically conductive and any cleanup methods should avoid contacting graphite with electrical circuitry.
- 6.1.2 For emergency responders: Wear approved dust mask, safety goggles, and conventional work gloves. Same methodology as for non-emergency personnel(sec 6.1.1)
- 6.2 Environmental Precautions: Synthetic graphite is inert and insoluble and will not pose any soluble ion hazards to the environment. However, good housekeeping practices should be followed and spilled material should be cleaned up, and disposed of in an appropriate manner.
- 6.3 Methods and material for containment and clean up: No special containment needed other than conventional vacuuming and waste containment. Avoid creating dust. Graphite is electrically conductive and any cleanup methods should avoid contacting graphite with electrical circuitry.
- 6.4 Reference to other sections: Not needed
- 6.5 Additional information: Not needed















Section 7 - Handling and Storage

7.1 Precautions for safe handling

7.1.1 Handling Use conventional methods, but avoid dusting conditions. Keep powder from contacting eyes. Synthetic graphite is a good conductor of electricity. Avoid contact between synthetic graphite and electrical circuitry.

Slip Hazard: Graphite is a highly lubricious material and may present a slip hazard if spilled on wet or dry pedestrian surfaces.

7.2 Conditions for safe storage, including any incompatibilities.

Storage and Incompatibilities Store all carbonaceous materials in a dry location. Graphite is incompatible with all oxidizing agents

Dust Explosibility Hazards: Very finely divided graphite powder poses a very slight risk of dust explosion hazard: Dust class ST1, MIE greater that 10 J (very low hazard of spark ignition)

Section 8 - Exposure Controls/ Personal Protection

8.1 Control parameters

orr control parameters						
8.1.1 Occupational exposure limits						
Component	CAS No.	%	ACGIH TWA	Control Reference		
Graphite	7782-42-5	100	2.0 mg/m ³ Respirable dust	2014 ACGIH TLV Handbook		
Graphite	7782-42-5	2-5 100 10.0 mg/m² F		2014 ACGIH TLV Handbook(insoluble particles not otherwise specified)		
Engineering Measures	Use adequate dust collection to maintain dust levels below the control or recommended values.					
Respiratory Protection	Approved dust mask, type N95 recommended.					
Eye Protection	Conventional safety glasses or goggles.					
Skin Protection	Conventional work gloves and clothing.					
Additional	Graphite spilled on pedestrian surfaces may pose a significant slip hazard.					

8.2 Exposure controls

- 8.2.1 Appropriate engineering controls: Use adequate dust collection to maintain dust levels below the control or recommended values.
- 8.2.2 Personal protective equipment
- 8.2.2.1 Eye/Face Protection: Wear laboratory goggles, or full side shielded safety glasses.
- 8.2.2.2 Skin Protection: Conventional work gloves and clothing.
- 8.2.2.3 Respiratory Protection: Approved dust mask, type N95 recommended.
- 8.2.3 Environmental exposure controls: Synthetic graphite is inert and insoluble. To the best of our knowledge, synthetic graphite should not present any environmental hazards. No special environmental exposure controls, other than standard practices for dust and spill control, are required.















Section 9 – Physical and Chemical Properties

9.1 Information on basic physical and chemical properties

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Color:	Gray to Black	Material State	Solid, granular or powder			
Odor	None					
Boiling Point:	NA	Melting Point	Sublimates at 3652C			
Specific Gravity	2.26	Vapor Density	Not applicable			
Vapor Pressure (mm Hg)	NA	% Volatile (By Wt.)	0-1%			
Solubility in Water	Insoluble	Evaporation Rate:	Not applicable			
рН	NA	Auto Ignition	Above 500 °C			
Decomposition Temp	Oxidizes above 450C	Dust Explosion class	ST1=KST>0-200 bar m/s, MIE			
			above 10 J.			
Flash Point	NA Solid substance with very high melting point.					

Section 10 - Stability and Reactivity

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10.1 Reactivity	Graphite is non-reactive under ambient conditions.
10.2 .Stability	Stable. Will not polymerize or self react spontaneously.
10.3 Possibility of	None known
hazardous reactions	
10.4 Conditions to Avoid	Avoid contact with oxidizing agents. Graphite will begin to oxidize at temperatures
	above 450 C.
10.5 Incompatible	Oxidizing agents
materials	
10.6 Hazardous	Carbon Dioxide (CO ₂), Carbon Monoxide (CO)
products of	
decomposition	
Flammable Limits	LEL and UEL values not available: Minimum Ignition Energy (MIE) greater than 10
(% by Vol.)	joules. When exposed to extremely high energy ignition sources very finely divided
	graphite powder can form explosive mixtures with air. Avoid contact between graphite
	dust clouds and high energy ignition sources. Classified as combustible but not
	flammable.

Section 11 – Toxicological Information

11.1 Information on toxicological effects: Acute toxicity

	Effect dose		,	Species	Method		Remarks	
Acute oral toxicity	LD50	LD50 > 2000 mg/kg bw			Rat	OECD 423		
Acute inhalation toxicity	LC5	0 > 2	000 mg/m3	00 mg/m3 R		OECD 403		Limit dose acc. to CLP.
		Spe	ecies	ecies Method				Result
Skin corrosion/irritation		Rab	bbit		OECD 404	1		Not irritating
Serious eye damage/irritation	on	Rab	bbit		OECD 405	5		Not irritating
Respiratory or skin sensitiza	ation	Мо	use		OECD 429)		Not sensitizing
	Speci	es	Method	Re	esult of effec	t dose	Remarks	
Genotoxicity	In vitr	O	OECD 471	Ne	egative	e Bacterial reve		everse mutation assay.
Genotoxicity	In vitr	O	OECD 473	Negative		Mammalian chromosome aberration		
					test.		test.	
Genotoxicity			OECT476 No		egative		Mammalian cell gene mutation test	
						(gene muta	ation).	
Carcinogenicity	Carcinogenicity		Literature	Not carcinogenic		nic	Based on available data the	
	(D		FG, 2002).		classification criteria are not me			
Reproductive toxicity	Rat	OECD 422 N		NC	NOAEL > 1000 Dose		Dose as nominal food intake,	
				mg	g/kg bw			ling to limit dose
							according t	to OECD 422. Based on
					ata the classification			
							criteria are	not met

















11.1 Information on toxicological effects: Acute toxicity continued

STOT-single exposure

Single exposure	Specific effect	Affected organs	Remark
Acute oral toxicity OECD 423 (rat)	No specific effects.	Not applicable.	Based on available data the classification criteria are not met.
Acute inhalation toxicity OECD 403 (rat)	Only usual signs of discomfort after the end of exposure were observed.	Not applicable.	Based on available data the classification criteria are not met.

STOT-repeated exposure

Repeated exposure	Specific effects	Affected organs	Remark
Sub-acute oral OECD 422 (rat)	No specific effects	Not applicable.	Based on available data the classification criteria are not met.
Sub-acute inhalation OECD 412 (rat)	Wet lung weight was increased. Minor histopathological findings in lung and nasal cavity	Respiratory tract.	Based on available data the classification criteria are not met.

Aspiration hazard: Solid substance. Based on available data the classification criteria are not met.

Symptoms related to the physical, chemical and toxicological characteristics

In case of ingestion: No signs of systemic toxicity found in studies acc. to OECD 423 and OECD 422.

No human data on effects after ingestion. See section 4 for first aid measures.

<u>In case of skin contact:</u> No irritation or corrosion found in a study acc. to OECD 404. No human data on effects after skin contact. See section 4 for first aid measures.

In case of inhalation: No signs of systemic toxicity found in studies acc. to OECD 403 and OECD 412.

Usual signs after inhalation of poorly soluble dusts with low toxicity were found in these studies. No symptoms are expected if relevant occupational exposure levels and derived no effect levels are complied with. In situations of repeated excessive lung overload due to a high airborne concentration of particles of respirable size for extended periods of time pneumoconiosis may develop. See section 4 for first aid measures.

<u>In case of eye contact:</u> No irritation or corrosion found in a study acc. to OECD 405. No human data on effects after eye contact. See section 4 for first aid measures.

11.2 Experiences made in practice

Observations relevant to classification: None. Other observations: None.

11.3 Other information

Neither signs for systemic toxicity nor for local skin-/eye-irritation nor sensitizing properties were found in any of the available studies. Repeated dose inhalation studies revealed some local effects generally observed after inhalation of poorly soluble dusts with low toxicity.















Section 12 - Ecological Information

12.1 Toxicity:	Synthetic graphite is inert and insoluble. To the best of our knowledge, synthetic does
	not present any significant environmental hazards

12.1.1 Aquatic Toxicity: Graphite is not water soluble and does not present a soluble-ion hazard. Fine graphite particles suspended in natural water bodies may be harmful to organisms sensitive to suspended solids.

Aquatic toxicity	Effect dose	Exposure time	Method	Remarks
Acute fish	LC50 > 100	96 hour	OECD 203	No adverse reaction up to the
toxicity	mg/l		(EU method C.1)	tested concentration could be observed.
Acute daphnia toxicity	EC50 > 100 mg/l	48 hour	OECD 202 (EU method C.2)	No adverse reaction up to the tested concentration could be observed.
Acute algae toxicity	EC50 > 100 mg/l	72 hour	OECD 201 (EU method C.3)	No adverse reaction up to the tested concentration could be observed.

12.1.2 Sediment toxicity: None known.

12.1.3 Terrestrial toxicity: None known.

- 12.2 Persistence and degradability: Graphite is a reduced form of carbon and will not degrade further under normal conditions. This form of carbon is stable, unreactive in water under ambient conditions, and is insoluble.
- 12.3 Bioaccumulation potential: There is no evidence indicating that graphite is bioaccumulative.
- 12.4 Soil Mobility: Graphite is not expected to have mobility in soil as it is an insoluble, inorganic substance.
- 12.5 PBT and vPvB assessment: Graphite is not a persistent bioaccumulative and toxic substance.
- 12.6 Other adverse effects: None known. Graphite has no ozone depleting potential.

Section 13 – Disposal Considerations

Dispose of in a manner which conforms to local, state and Federal regulations.

Graphite is a reduced form of carbon. Graphite is non-hazardous but disposal of graphite waste should be handled in a responsible matter. Dust formation from packaging residues should be avoided. Store empty packaging in a suitable receptacle.

Graphite is a form of elemental carbon so it is not biodegradable.

Provision of a European Waste Catalog, waste code number, should be handled in agreement with the regional waste disposal company.

Packaging should be completely emptied of contents and disposed of in a manner specified by the recycler/regional disposal contractor. Dust formation from packaging residues should be avoided. Store empty packaging in a suitable receptacle.

Section 14 - Transport Information

14.1 UN Number	Not applicable
14.2 UN Proper shipping name	Not applicable
14.3 Transport hazard class	Not applicable
14. 4 Packing Group	Not applicable
14.5 Environmental hazards	None known
Marine Transport	Not classified as a hazardous material
Land Transport	Not classified as a hazardous material
Air Transport	Not classified as a hazardous material
Transport Label Required	No label required

















Section 15 – Regulatory Information

15.1 Regulatory Status and Inventories

13.1 Negulatory Status and inventories		
Not Classified		
Inventory Information:		
EEC EINECS	#231-955-3	
US TSCA	Yes	
Canada DSL	Yes	
Canada NDSL	No	
Australian AICS	Yes	
Korean ECL	Yes	
Asia PAC	Yes	
Swiss Giftliste 1	Yes #G8422	
IECSC	Yes	
PICCS	Yes	
New Zealand NZLoC	Yes	
REACH: Fully registered		
RoHS: Synthetic graphite is compliant with the EU RoHS directive		
WEEE: Synthetic graphite is compliant with the EU waste electrical and electronic equipment directive		
15.2 Chemical Safety Assessment: For this substance a chemical safety assessment is not required		

Section 16 - Other Information

Abbreviations Used:

ACGIH TWA American Council of Government and Industrial Hygienists Time Weighted Average value.

CAS Chemical Abstracts Service

NA Not applicable

N.O.S. Not otherwise specified













